Petrology and Geochemistry of lavas from Sal Island: Implications for the variability of the Cape Verde magmatism.

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Keywords: Cape Verde Archipelago; Mantle plume; EM1; HIMU; Sal Island.

Abstract: Sal Island (Cape Verde Archipelago) preserves a magmatic history ranging from ≈ 25.6 Ma to ≈ 0.6 Ma. The chemistry of magmatism varies from Hy-normative (Serra Negra basalts: ≈ 5 Ma) to highly SiO₂-undersaturated (nephelinites and melilities). The calculated temperatures of magmatic segregation are higher than those usually considered for the asthenosphere, thus endorsing the role of a mantle plume to the origin of Cape Verde magmatism. Nephelinites and melilities are characterized by pronounced K negative anomalies which are interpreted as reflecting the interaction of plume magmas with the lithosphere. Some of the Sal lavas are characterized by low ¹⁴³Nd/¹⁴⁴Nd (down to 0.512577) – when compared with the HIMU component. These values are similar to those usually reported for the Southern Cape Verde Islands. This suggest the role of a EM1-type end-member to the magma sources of the Sal, which is for the first time evidenced from the whole rock composition of one of the Northern Cape Verde Islands.

Palavras-chave: Arquipélago de Cabo Verde, Pluma mantélica, EM1, HIMU; Ilha do Sal.

Resumo: A ilha do Sal (Arquipélago de Cabo Verde) preserva uma história magmática estendendo-se desde os ≈ 25.6 Ma aos ≈ 0.6 Ma. As características químicas do magmatismo variam desde composições caracterizadas por hiperstena normativa, em rochas com cerca de 5 Ma (Formação da Serra Negra), até litótipos altamente subsaturados em sílica (nefelinitos e melilititos). As temperaturas de extracção magmática calculadas indicam valores superiores aos normalmente referenciados para a astenosfera, suportando os modelos que invocam o papel de uma pluma mantélica para a origem dos magmas de Cabo Verde. Nefelinitos e melilititos caracterizam-se por pronunciadas anomalias negativas em K que são interpretadas como resultado da interacção dos magmas provenientes da pluma mantélica com a litosfera. Algumas das lavas da ilha do Sal caracterizam-se por valores de 143 Nd/¹⁴⁴Nd baixos (até 0.512577) quando comparados com os que caracterizam o componente HIMU. Tais características são similares às que têm sido descritas para as ilhas de Sul. Tal indica que, em oposição ao normalmente considerado, a contribuição do componente EM1 para as fontes mantélicas de Cabo Verde não se restringe às ilhas do Sul.

1 – INTRODUCTION

Cape Verde archipelago $(14^{\circ}48' - 17^{\circ}12'N; 22^{\circ}44' - 25^{\circ}28'W)$ is located on the East Central Atlantic between the Equator and the Cancer Tropic, some 600 to 900 km away from the western African coast and ≈ 2000 km to the east of the Mid Atlantic Ridge.

The archipelago (Fig. 1) is composed of two groups of islands (windward vs. leeward or Northern vs. Southern), a subdivision that also reflects their different geochemical characteristics. Southern Islands are less Nd- and Pb-radiogenic and more Sr-radiogenic than the Northern Islands, which are also characterised by more unradiogenic He signatures. Such distinction is also

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evident from a ²⁰⁶Pb/²⁰⁴Pb vs. ²⁰⁸Pb/²⁰⁴Pb diagram. Whereas the Northern Islands plot along the North Hemisphere Reference Line (NHRL; HART, 1984) the Southern Islands are mainly positioned above the NHRL. This dichotomy is considered to reflect the distinct contribution of HIMU- and EM1-type mantle end-members to the sources of these two groups of islands. On Southern Islands an EM1-type end-member dominates whereas on the Northern Islands the isotopic signatures are dominated by a HIMU-type end-member, flavoured with a contribution of the lower mantle in order to account for the unradiogenic He isotopic signatures. The additional contribution of a depleted end-member is mainly required to explain the composition of Northern Islands (GERLACH et al., 1988; DAVIES et al., 1989; CHRISTENSEN et al., 2001; JØRGENSEN & HOLM, 2002; DOUCELANCE et al., 2003; ESCRIG et al., 2005; HOLM et al., 2006; MILLET et al., 2008; BARKER et al., 2009a, b; MARTINS et al., 2010). For submarine lavas of the Flamengos formation (Santiago Island) it was also proposed the contribution of a FOZO-like component (BARKER et al., 2009a; b).

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We would like to emphasize that such geochemical distinction between Northern and Southern Islands has been made based upon the study of an incomplete set of data. Indeed, and particularly in what concerns Sal Island, the available isotope data is limited to ⁸⁷Sr/⁸⁶Sr and Pb isotopic ratios for one and three samples, respectively (DOUCELANCE *et al.* 2003).

In this paper we will present detailed geochemical data from Sal Island lavas, which will be discussed in conjunction with the available cartographic (SILVA *et al.*, 1990) and geochronological (TORRES *et al.*, 2002a; HOLM *et al.*, 2008) data.

2 - GEOLOGICAL SETTING

The Cape Verde archipelago occurs on an ≈ 83 km thick lithosphere (CAZENAVE *et al.*, 1988) with 120 to 140 Ma (WILLIAMS *et al.*, 1990; MÜLLER *et al.* 2008) and is composed by 10 islands and 8 islets presenting a horse-shoe geographical distribution opening westward. It is



Fig. 1 - Location of Sal Island at the Cape Verde Archipelago.

located on the southwest part of the Cape Verde Rise, a ≈ 2.2 km high and 1400-1600 km wide swell considered the largest oceanic intraplate bathymetric anomaly (e.g. LODGE & HELFFRICH, 2006). Crustal thickness is anomalously high (up to 22 km) beneath the islands, which is attributed to magmatic transfer from the mantle plume source to crustal levels (LODGE & HELFFRICH, 2006). However, between the islands the crust is only ≈ 7 km thick, which is normal for oceanic settings (ALI *et al.*, 2003; PIM *et al.*, 2008).

The positioning of the Cape Verde archipelago is also coincident with important geoid (7.6 \pm 0.3 m) and heat flow (16 \pm 4 mWm⁻²) anomalies (COURTNEY & WHITE, 1986; MONNEREAU & CAZENAVE, 1990) being characterized by a geoid height to topography ratio (4.4 \pm 0.23) consistent with the reheating of the lower lithosphere (GREVEMEYER, 1999). The Cape Verde islands are flanked by a flexural moat, which is less important than it would be expected from surface volcanic(s) loading. This was interpreted as suggesting the existence of subsurface, upward acting, loads counteracting that surface downward loading (ALI *et al.*, 2003). All these suggest the role of a mantle plume to the genesis of Cape Verde.

The genesis of Cape Verde from a deeply anchored mantle plume seems to be supported by seismic tomography studies imaging a low S-wave velocity domain extending down to 2800 km (MONTELLI et al., 2006; see also ZHAO, 2007) and by unradiogenic He isotope signatures obtained both from silicate and carbonatitic rocks (R/Ra up to 15), which point to the contribution of a high ³He/⁴He reservoir, plausibly located in the deepest parts of the lower mantle (cf. CHRISTENSEN et al., 2001; DOUCELANCE et al., 2003; MOURÃO et al., 2007; MATA et al., 2010). However, a recent analysis of P-to-S receiver functions showed that the time separation between the 410 and 660 km discontinuities is not modified beneath Cape Verde (HELFFRICH et al., 2010), suggesting that the plume role on the magma genesis is still an open question in Cape Verde.

The region has been subjected to very important vertical movements testified by the occurrence of MORB at Maio and Santiago islands (DE PAEPE *et al.*, 1974; GERLACH *et al.*, 1988; MILLET *et al.*, 2008), by seafloor sediments of disputable Lower Cretaceous age cropping out in Maio (e.g. AZÉMA *et al.*, 1990; HOLM *et al.*, 2008) and by the occurrence on several islands of uplifted submarine alkaline lavas, which, for example, reach 450 m apsl in Santiago Island and 400 m apsl in Brava Island (SERRALHEIRO, 1976; ZAZO *et al.*, 2007; HOLM *et al.*,

2008; RAMALHO *et al.*, 2010 a; b; MADEIRA *et al.*, 2008; 2010; MOURÃO *et al.*, 2010).

3 - GEOLOGY AND GEOCHRONOLOGY

With 216 km², Sal is an arid, highly eroded and flat island reaching the highest height at Monte Grande (406 m), a Quaternary lapilli cone.

Extensive field work and K-Ar determinations (TORRES *et al.* 2002a, see Figs. 2 and 3) enabled the proposition of the following volcanostratigraphic subdivisions (see also TORRES *et al.*, 2002b and HOLM *et al.*, 2008).

The Old Eruptive Complex is mainly composed by submarine lavas and hialoclastites of alkali basaltic composition. The unaltered core of a pillow lava allowed the determination of a reliable K-Ar age of 25.6 Ma (TORRES *et al.*, 2002a) corresponding to the most ancient isotopic age obtained for the Cape Verde hotspot. These rocks were intruded by an impressive dyke swarm of variable composition (basaltic to trachytic) with N70°E, E-W and N20°W as dominant strikes. N70°E is the dominant orientation to the north of the Leteu rivulet, where dykes correspond to about 80 % of the outcropping area. N20°W is the dominant dyke orientation in the region between Madama de Baixo and Madama de Cima rivulets (Fig. 2).

Intrusives bodies (foidic syenites, gabbros and alkali pyroxenites) which sometimes occur as zoned subcircular complexes, are interpreted as resulting from the solidification of subvolcanic magma chambers. In these subsuperficial reservoirs basic *Ne*-normative magmas gave origin to pyroxenitic cumulates and chemically evolved to produce (theralitic) gabbros, essexites and nepheline syenites (SILVA, 1991, CALDEIRA & SILVA, 2008).

Micritic limestones included in the volcanic products of this complex (SILVA *et al.* 1990; UBALDO *et al.* 1991) attests to the submarine origin of part of the sequence. Furthermore, according to RAMALHO *et al.* (2010 a), the high planktic/benthic ratio indicates pelagic conditions associated with depths in excess of 300-400 m, suggesting that the Old Eruptive Complex corresponds to the seamount stage of the Sal development and implying a significant (> 350 m) uplift of the volcanic island.

Subaerial volcanism is testified by some rare and volumetrically unimportant flows and dykes of melanephelinites and by flows, dykes and chimneys of phonolitic composition.



Fig. 2 – Geological map of Sal Island showing the stratigraphic units (after TORRES *et al.*, 2002a), simplified and adapted from "Carta Geológica da ilha do Sal, na escala 1/25 000" (SILVA *et al.*, 1990).



Fig. 3 – Illustration of the available age determinations for Sal Island (TORRES *et al.*, 2002a; HOLM *et al.*, 2008). OEC – Old Eruptive Complex, MEF – Main Eruptive Formation, PAB – Ponta Altar Baleia Complex, SN – Serra Negra Formation, MGPL – Monte Grande Pedra Lume Formation. Black question mark indicates that the age obtained for the essexitic rock of a subcircular complex, thought to be part of the OEC, is close to the age determined for basalts from the MEF. This raises the possibility that some of the intrusives in OEC are coeval with the overlying volcanic sequences, as suggested by TORRES *et al.* (2002a). However, field evidences and a possible age resetting must be taken in consideration. Hence the doubt about the statigraphic position of these intrusives remains until further age determinations are obtained.

After a period of intense dissection, the eruption rates reached its highest values during the genesis of the **Main Eruptive Formation** when submarine and subaerial alkali olivine basalts (15.8 Ma, TORRES *et al.*, 2002a) and olivine nephelinites (14.2 Ma; HOLM *et al.*, 2008) were formed. This sequence, mainly outcropping in the Southeastern part of the island is followed by the **Ponta do Altar-Baleia Complex**. In this complex, a marine deposit (\approx 2m thick), including calcarenites, a conglomeratic level and, towards the top, siltites and argillites, separates the underlying alkali basaltic rocks from the olivine nephelinites. For the upper nephelinitic rocks, outcropping at Morrinho do Açúcar volcanic chimney, ages of 8.7 Ma (TORRES *et al.*, 2002a) and of 11.2 Ma (HOLM *et al.*, 2008) were obtained. It is worthwhile to notice that TORRES *et al.* (2002a) obtained 9.7 Ma for a nephelinite outcropping in the Serra Negra region, which was considered, by field arguments, to be part of the Main Eruptive Formation (upper sequence). This formation also outcrops on the southeastern extreme of the island where HOLM *et al.* (2008) obtained the 40 Ar/ 39 Ar age of 14.2 Ma. Such significant age discrepancy and the fact that the 9.7 Ma age was obtained from a nephelinite cut by latter ijolitic pegmatitoids, advise for the necessity of an age confirmation for the Main Eruptive Formation nephelinites.

After the formation of these nephelinitic rocks, volcanic activity decreased significantly or even stopped. During this period of volcanic quiescence erosion was the main shaping process of the island being testified by the formation of a lahar type deposit. During this period marine limestones and calcarenites were also deposited. The study of planctonic foraminiferous allows considering the limestones of an Upper Miocene age (M. Lourdes Ubaldo, oral com.) which conforms with the radiometric ages of the enclosing volcanic rocks. The Main Eruptive Formation is cut by a prominent dyke swarm of a remarkably constant orientation (N40°W) suggestive of the existence of a clearly anisotropic stress field with σ 1 (compression) oriented to N40°W.

A new period of volcanic activity occurred by the end of Miocene with the genesis of the **Serra Negra Eruptive Formation** (~ 5.5 Ma; TORRES *et al.*, 2002a; HOLM *et al.*, 2008) in which submarine volcanic facies are gradually replaced by subaerial ones. This is the only period marked by the formation of Hy-normative basaltic rocks at the Sal Island. In opposition with the dominant fissural volcanism during the development of the Old Eruptive Complex and the Main Eruptive Formation, volcanic activity is now clearly centered.

A long period of volcanic quiescence (≈ 5 Ma) occurred until the upper Pleistocene when highly undersaturated rocks (olivine nephelinites and olivine melilitites) were erupted (lava flows and pyroclastic cones) to give origin to the Monte Grande-Pedra Lume Formation (1.06 to 0.6 Ma, HOLM et al., 2008; TORRES et al., 2002a). An 1100 m diameter crater generated by a phreatic/phreatomagmatic explosion (SILVA et al., 2002) is the most significant geomorphological expression of this formation; the evaporitic salt deposits continuously forming within it are clearly related with the name of the Sal island. The cones from which the Pedra Lume volcanics erupted occupy a belt limited by the fault zone of Madama de Baixo rivulet. This belt, with a orientation of NE - SW, probably extend seaward as it is suggested by the submarine apron evident from bathymetry of the NNE extremity of the island.

The highly eroded and low-lying morphology of Sal is dominated by extensive wave-cut platforms and terraces mostly formed during Quaternary (BEBIANO 1932; LECOINTRE 1963; SERRALHEIRO 1968; SILVA *et al.*, 1990; TORRES *et al.*, 2002b, ZAZO *et al.*, 2007). Recently ZAZO *et al.*, (2007) identified 16 terrace levels from 1 to 60 m a.s.l. but, as noticed by RAMALHO *et al.* (2010a), omitted the 100 m level, on top of Serra Negra, previously identified by SILVA *et al.* (1990). According to RAMALHO *et al.* (2010a), this terrace is older than 1 Ma.

4 – ANALYTICAL METHODS

A subset of 58 samples were selected for analysis from a larger collection taking in account the defined volcanostratigraphy and a careful thin section study (which screened out samples presenting strong evidences of meteoric/hydrothermal alteration and/or where phenocrysts represent > 10% volume). These rocks were analyzed for major and trace elements (Cr, Ni, Cu, Zn, Rb, Ba, Sr, Nb, Zr, and Y) at the CCM – IICT (Centro de Cristalografia e Mineralogia do Instituto de Investigação Científica Tropical. Portugal) by X-ray fluorescence spectrometry (Philips PW 1400). 18 samples were further analyzed for REE (and Sc, Cs, Ta, Hf, and U) by neutron activation (INAA) at the Activation Laboratories LTD (Ontario, Canada) and for 87Sr/86Sr and 143Nd/144Nd at the CPGeo-IGUSP (Centro de Pesquisas Geocronológicas do Instituto de Geociências da Universidade de São Paulo, Brasil).

Sm-Nd analytical procedures followed those described by SATO *et al.* (1995).

Mass spectrometric analyses were performed on a VG ISOMASS 354 fully automated mass spectrometer with five collectors. Sm and Nd analysis were determined by isotopic dilution technique, using a mixed ¹⁴⁹Sm-¹⁵⁰Nd spike, with an Nd concentration of 0.0003314 mmol/g and ¹⁴⁹Sm content of 0.004575 mmol/g. Nd and Sm were loaded as phosphates on two Ta side filaments of a triple filament assembly, with Re as the ionizing central filament. The quality of the analytical measurements by isotopic dilution techniques was evaluated using a standard ¹⁴²Nd/¹⁴⁴Nd ratio of 1.141828. The Nd isotopic compositions were corrected for mass fractionation by normalizing to a value of 0.7219 for ¹⁴⁶Nd/¹⁴⁴Nd.

Average ¹⁴³Nd/¹⁴⁴Nd for La Jolla and BCR-1 standards, yielded results of 0.511847 \pm 0.000022 and 0.512662 \pm 0.000027, respectively.

Rb and Sr contents and Rb/Sr ratios were determined by X-Ray fluorescence on powders by means of a Philips PW- 1380/00 automatic spectrometer. Samples with low Rb and Sr contents (2ppm), were re-analyzed by isotopic dilution technique, using ⁸⁵Rb and ⁸⁴Sr spikes. The samples were dissolved with HF+HNO₃ overnight at 65°C, and the Sr purified using a conventional ion exchange column with AG50WX8 cation exchange resin (200-400#) as described, with some modifications, by KAWASHITA (1972). The Sr was loaded on a single Ta filament with phosphoric acid. The ⁸⁷Sr/⁸⁶Sr ratios were corrected for mass fractionation by normalizing to a value of 0.1194 for ⁸⁶Sr/ ⁸⁸Sr. During the period of this work, six analyses of the NBS-987 standard averaged 0.71026 ± 0.00002 (Is). Procedural blanks for Sr were of the order of 1 ng. Errors reported here for initial ⁸⁷Sr/⁸⁶Sr ratios (ISr) are at the 2s level.

Mineral compositions were determined at Centro de Geologia da Universidade de Lisboa (CeGUL), on polished thin sections, using a 5 µm beam in a three channel wavelength dispersive JEOL-JCXA733 electron microprobe, operated at an accelerating voltage of 15 kv and a probe current of 25nA. Standards were natural homogeneous minerals (Si-wollastonite and olivine; Ca-wollastonite; Ti and Na - kaersutite; Ti - ilmenite; Al-disthene; Fe and Mg - olivine; Mn-rhodonite; k-adularia) and syntetic minerals of Cr, V and Ni. Precision, as indicated by replicate determinations on an in-house standard, is better than 2% for major elements. The mineral chemistry data base was recently slightly enlarged with the addition of few analyses made at CeGUL on a JEOL JXA 8200 microprobe, equipped with 4 wave-length dispersive spectrometers (WDS) and an energy dispersive detector (EDS), using a range of natural minerals as standards. An accelerating voltage of 15 kV was combined with a beam current of 10 nA and a beam diameter of 5 µm.

5 - PETROGRAPHY AND MINERAL CHEMISTRY

5.1 Petrography

The main lithotypes representing Sal island lavas are alkali basalts, nephelinites and melilitites.

Alkali basalts are hypocrystaline to holocrystaline, mainly porphyric but occasionally aphyric rocks, frequently with vesicles filled with zeolites or carbonates. Olivine is the main phenocryst phase followed by subordinate clinopyroxene. Plagioclase is rare and amphibole occurs as a phenocryst phase in basalts from the Main Eruptive Formation. Euhedral to subeuhedral olivine (Fo₆₂₋₈₇), up to 8 mm across, often displays alteration to iddingsite and/or iron oxides/hydroxides along cracks. Occasionally they exhibit skeletal morphology indicating high growing rates stemming from the swiftness of quenching processes. Purple-brownish Ti-augite $(< 1.5 \text{ mm across; } Wo_{48-52}En_{36-43}Fs_{9-12})$ frequently display zoning marked by a core to rim Ti enrichment (e.g. TiO₂ from 1.79 to 2.42 wt%). Hourglass-type zoning is occasionally present. Plagioclase, when

present, may reach up to 1.5 mm across, and show corerim zoning from An₇₃ to An₅₇. Amphibole is anhedric, pleochroic, frequently with reabsorbed rims where opaque minerals dominate. This is interpreted as the result of destabilization of amphibole due to devolatilization affecting magma during ascent. The groundmass in alkali basalts invariably consists of plagioclase laths (65.21 – 48.84 mol % An), occasionally oriented, Ti-augite (Wo₄₈₋₅₀En₃₃₋₄₁Fs₁₁₋₁₇; TiO₂ \approx 5.19 wt%), and opaque oxide grains. Olivine may also occur. Interstitial brown isotropic glass is present in the groundmass of submarine lavas. Serra Negra Formation hyaloclastites locally show signs of phosphatization, demonstrated by the presence of phosphate minerals predominantly of the variscite type (SILVA *et al.*, 1993).

Nephelinites and melilitites are holocrystalline, porphyric, sometimes with glomeroporphyritic texture. The phenocrystal assemblage comprises olivine ($Fo_{81,87}$), clinopyroxene and opaque minerals. Olivine (up to 5 mm), sometimes with primary fluid inclusions, is frequently altered to iron oxides/hydroxides and serpentine. Some olivines display kink-bands (indicating deformation inside a solid matrix), allowing the interpretation that they correspond xenocrystals derived from xenoliths desegregation. The main clinopyroxene type is Ti-augite $(Wo_{51-54}En_{31-40}Fs_{10-15}, TiO_2 = 6.33 - 7.25 \text{ wt \%})$, up to 2.5 mm across and sometimes with hourglass zoning. Ti-augites are sometimes totally or partially replaced by crystal aggregates of yellowish-green esseneites $(Wo_{57-70}En_{29-36}Fs_{7-0.7}; TiO_2 = 0.23 - 1.99 \text{ wt \%})$ that were interpreted as result of subsolidus fluid/rock interaction under high oxygen fugacity conditions (MUNHÁ et al., 1991). The groundmass includes olivine, variable amounts of clinopyroxene (Wo₅₁₋₅₄En₃₄₋₃₉Fs₉₋₁₂), nepheline (sometimes in interstitial poeckilitic crystals), melilite (Aker₆₆Gehl₃₄; alongated microliths frequently altered and displaying typical peg-structure), perowskite, biotite/phlogopite and Fe-Ti oxides. Some samples have ocelli-like vesicles filled with carbonates and zeolites. Flakes of dark-red rhönite and small crystals of hauyne are present in the matrix of some olivine melilitites from the Monte Grande - Pedra Lume Formation. Variable proportions of clinopyroxene, nepheline and melilite in the groundmass lead to transitions between olivine melanephelinites, olivine nephelinites with melilite, and olivine melilitites.

Mantle xenoliths such as dunite/wherlite cumulates and lherzolite/harzburgite refractory residues are present in nephelinitic and alkali basaltic lavas (DE PAEPE & KLERKX, 1971; KOGARGO & SENIN, 1993; RYABCHIKOV *et al.*, 1995; BONADIMAN *et al.*, 2005; MENDES *et al.*, 2005; SHAW *et al.*, 2006). Gabbro and gabbronorite xenoliths included in olivine melanephelinites from Monte Grande – Pedra Lume Formation were interpreted as representing fragmental pieces of the oceanic crust crossed by ascending magmas (MENDES & SILVA, 2001).

5.2 Mineral Chemistry

Representative electron micropobe analyses of phases in the volcanic rocks from Sal Island are given in tables 1 - 6.

Olivine is compositionally variable with Fo contents ranging from 62 to 87 mol %, MnO from 0.11 to 0.77 wt%, CaO from 0.15 to 1.01 wt %, and NiO from 0.01 to 0.45 wt%. Olivine in nephelinitic/melilitic rocks tends to have lower NiO concentration at a given Fo, when compared to basaltic olivines. Phenocrysts are usually zoned, with Fo and Ni decreasing and Ca and Mn increasing from core to rim. The increase of Ca content normally is attributed to a response to pressure decrease during magma ascent (HIRSCHMANN & GHIORSO, 1994). However, the distinct evolutionary trends displayed by olivine in basaltic and nephelinitic rocks (Fig. 4) show that the composition and crystallization temperature. Indeed it has been suggested that CaO content of olivines varies inversely with silica activity whereas $D_{N_i}^{ol/liq}$ increases with the decrease of polymerization degree of the magma (e.g. WATSON, 1979; MATA, 1996; 1998). This is supported by the Sal data depicted in Fig. 6 taking into account that SiO₂ range from 39 to 45 wt% in basalts and from 34 to 40 wt % in nephelinites (see table 7).

Estimation of olivine pnenocrysts crystallization temperatures by the methods of ROEDDER & EMSLIE (1970) and BEATTIE (1993), indicates values from 1266 to 1326 °C for basaltic rocks and 1324 to 1363 °C for nephelinites/melilitites. For these calculations only olivine-whole rock pairs for which $KD_{Fe/Mg}^{\alpha//Mg} \approx 0.30$ were used in order to ensure that crystal/liquid equilibrium was attained (ROEDDER & EMSLIE, 1970).

Most **pyroxenes** found in Sal Island basaltic and nephelinitic lavas are classified as diopsides although in many of the nephelinitic/melilitic rocks they have Wo > 50 mol %, as it is typical for silica-undersaturated lavas. Some of the pyroxenes in nephelinites/melilitites are esseneites resulting from low-temperature recrystallization of diopsides (MUNHÁ *et al.*, 1991). Diopsidic clinopyroxenes are enriched in Al₂O₃ and TiO₂ (up to 10.85 and 6.65 wt % respectively) with the highest concentrations confined to the rims and groundmass. Diopsides are Ca rich (CaO = 21 to 25 wt %) showing a general increase from alkali basalts (Ca < 0.94 a.f.u.) to

TABLE 1	
Representative olivine microprobe analyses of Sal lavas	

Geol Unit	Main	Eruptive	Format	ion		Por	nta do Al	tar Balei	а			Se	rra Negr	a			Monte (Grande -	Pedra L	ume	
Lithotype	Basanit	e	Nephel	inite	Basanit	е	Basanit	e	Nephel	inite	Basanit	e	Basanit	te		Basanit	te	Nepheli	inite	Nepheli	nite
Sample	115		116		679		1001		424		1025		109			682		238		1315	
	С	R	С	R	С	R	С	R	С	R	С	R	С		R	С	R	С	R	С	R
SiO ₂ (wt %)	37.08	36.55	39.80	39.54	39.70	39.20	39.96	39.36	40.15	39.58	40.44	39.28	40.25	39.77	39.64	39.89	38.80	40.40	39.80	40.31	39.73
TiO₂	0.06	0.06	0.04	0.00	0.00	0.06	0.02	0.01	0.05	0.02	0.01	0.02	0.03	0.01	0.03	0.03	0.07	0.04	0.03	0.05	0.04
Al ₂ O ₃	0.02	0.04	0.04	0.00	0.04	0.00	0.05	0.03	0.05	0.02	0.02	0.04	0.03	0.02	0.04	0.05	0.05	0.06	0.02	0.03	0.03
NiO	0.06	0.03	0.11	0.10	0.20	0.14	0.25	0.15	0.24	0.04	0.27	0.18	0.28	0.29	0.21	0.27	0.10	0.23	0.14	0.26	0.08
FeO	29.56	32.15	14.79	15.98	14.51	19.46	15.24	16.35	12.93	14. 10	13.59	16.48	12.96	13.34	14.45	13.58	21.63	12.99	13.50	12.42	13.66
MnO	0.66	0.77	0.33	0.42	0.21	0.24	0.20	0.25	0.20	0.49	0.18	0.28	0.18	0.19	0.19	0.21	0.33	0.17	0.22	0.17	0.39
MgO	31.80	29.72	44.02	43.35	44.68	40.33	44.29	43.37	46.54	43.78	45.86	42.73	46.58	46.29	44.63	45.60	38.33	46.02	45.66	46.74	44.96
CaO	0.45	0.53	0.46	0.69	0.25	0.36	0.22	0.36	0.19	0.99	0.22	0.29	0.20	0.23	0.34	0.24	0.49	0.13	0.25	0.15	0.73
Total	99.68	99.85	99.59	100.08	99.60	99.79	100.23	99.89	100.35	99.03	100.60	99.29	100.50	100.15	99.53	99.87	99.79	100.03	99.62	100.12	99.62
% Fo	65.73	62.24	84.14	82.87	84.59	78.69	83.82	82.54	86.52	84.70	85.75	82.22	86.50	86.09	84.63	85.68	75.96	86.33	85.77	87.03	85.43
% Fa	34.27	37.76	15.86	17.13	15.41	21.31	16.18	17.46	13.48	15.30	14.25	17.78	13.50	13.91	15.37	14.32	24.04	13.67	14.23	12.97	14.57
Si ⁺⁴	1 006	1 003	1 004	0.999	1 000	1 008	1 003	0.998	0.996	1 003	1 003	1 002	0.997	0.992	0,999	0.998	1 008	1 004	0.997	0.999	0.998
Ti *4	0.001	0.001	0.001	0.000	0.000	0.001	0.000	0.000	0.001	0.000	0.000	0.000	0.001	0.000	0.001	0.000	0.001	0.001	0.001	0.001	0.001
AI * ³	0.001	0.001	0.001	0.000	0.001	0.000	0.002	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.002	0.002	0.000	0.001	0.001
Ni⁺²	0.001	0.001	0.002	0.002	0.004	0.003	0.005	0.003	0.005	0.001	0.005	0.004	0.006	0.006	0.004	0.005	0.002	0.005	0.003	0.005	0.002
Fe⁺²	0.670	0.738	0.312	0.338	0.306	0.418	0.320	0.347	0.268	0.299	0.282	0.351	0.269	0.278	0.304	0.284	0.470	0.270	0.283	0.257	0.287
Mn ⁺²	0.015	0.018	0.007	0.009	0.004	0.005	0.004	0.005	0.004	0.011	0.004	0.006	0.004	0.004	0.004	0.005	0.007	0.004	0.005	0.004	0.008
Mg *²	1.286	1.217	1.655	1.634	1.677	1.546	1.657	1.638	1.721	1.655	1.696	1.625	1.720	1.721	1.677	1.701	1.485	1.706	1.706	1.728	1.684
Ca⁺²	0.013	0.016	0.013	0.019	0.007	0.010	0.006	0.010	0.005	0.027	0.006	0.008	0.005	0.006	0.009	0.006	0.014	0.004	0.007	0.004	0.020

C – core, R – rim, I – intermediate zone. For the complete set of analyses see Torres (1998).

More interplace in the interplace int	Geol Uhit	2	1 ain Eruptiv	ve Form atic	F				Sena Negr	-				Pon	a do Altar E	saleia							Monte	Srande Ped	a-Lume					
Mare Mare <th< th=""><th>Lithotype</th><th>Basalt</th><th></th><th></th><th></th><th></th><th>3asalt</th><th></th><th>Ba</th><th>salt</th><th></th><th>Ba</th><th>salt</th><th></th><th></th><th>Basalt</th><th></th><th></th><th>Basalt</th><th></th><th></th><th></th><th>Nephelinit</th><th></th><th></th><th>Ż</th><th>N yde</th><th>phelinite</th><th></th><th></th></th<>	Lithotype	Basalt					3asalt		Ba	salt		Ba	salt			Basalt			Basalt				Nephelinit			Ż	N yde	phelinite		
C I R M C I M C I M C I M C I M C I M C I M C I M	Sample	115				~	025		10			675				1001			682				116			12	48* 12	47*		
00.1 013 <th></th> <th>υ</th> <th></th> <th>ľ</th> <th>2</th> <th>_</th> <th>-</th> <th>Σ</th> <th>ບ _</th> <th> -</th> <th>Σ</th> <th>υ</th> <th> -</th> <th>¥</th> <th>Σ</th> <th>U</th> <th>¥</th> <th>Σ</th> <th>υ</th> <th>_</th> <th>¥</th> <th>Σ</th> <th>υ</th> <th></th> <th>2</th> <th>-</th> <th>-</th> <th>-</th> <th>-</th> <th>Ĩ</th>		υ		ľ	2	_	-	Σ	ບ _	-	Σ	υ	-	¥	Σ	U	¥	Σ	υ	_	¥	Σ	υ		2	-	-	-	-	Ĩ
M01C2C3C4 <th></th>																														
Display 123 143 123 143 123 143 123 143 123 143 123 143 123 143 123 143 123 143 123 143 123 143 123	SiO ₂ wt%	50.32	48.92	47.53	46.39	44.76	48.39	48.73	44.72	46.36 4	7 62.74	45.18	19.59 50	.66 49.	21 49.16	\$ 49.8	4 49.25	5 46.63	43.61	46.29	47.93	48.30	40.31	42.89	43.59	47.38	40.45	50.43 4	1.27 38	.26
A,0i = 3i	T0 2	1.23	1.49	2.25	2.51	3.74	2.15	2.06	4.37	3.12	251	3.29	1.79	.55 2.	42 1.8	3 1.6	34 1.6	1 3.20	3.99	2.79	2.88	2.72	5.83	4,44	4.01	2.89	6.65	234	1.25 C	23
Cy0110:211:121:013:023:013:023:013:023	AIJO	4.59	5.41	6.41	7.20	7.43	4.30	4.05	6.75	6.57	6.07	6.73	5.13 3	.79 4.	57 4.9	0 4.2	21 4.4	2 5.26	9.73	7.26	5.38	4.59	10.28	8.56	8.09	4.68	10.85	287	.62 13	74
He-Performentiality in the second of the	cr203	0.22	0.31	0.25	0.20	0.04	0.08	0.16	0.00	0.49	0.30	0.03	0.56 0	.43 0.1	33 0.3	1 0.3	52 O.3	4 0.00	0.52	0.40	0.00	0.00	0.00	0.00	0.04	0.03	0.05	.p.u	.01 n.	71
Model314	Fe ₂ 0 ₃ #	1.69	2.67	3.13	2.98	3.06	2.21	2.51	3.61	1.56	226	4.00	0.73 0	.48 1.	48 2.1	4 0.5	50 1.1	3 1.43	2.08	2.99	0.35	0.67	4.57	4.08	5.34	4.01	4.20	0.43 1	15 15	31
Mode1000000000010010010000010000010000010000010000010000010000010000010	FeO #	3.84	3.28	3.14	3.77	5.94	5.55	5.06	4.84	5.47	4.79	4.83	5.43 5	47 6.	74 4.3	8 5.7	3 5.1	1 10.21	5.05	3.92	8.40	9.28	2.63	2.69	1.58	1.98	3.20	4.79	250 C	32
MCDMC	QLM	0.08	0.06	0.06	0.07	0.11	0.13	0.13	0.15	0.11	0.09	0.17	0.11 0	12 0.	15 0.1	0.0	9 0.1	1 0.16	0.05	0.08	0.18	0.22	0.06	0.04	0.03	0.08	0.09	0.10	0.18 C	17
Ccd13123123623	QGM	14.82	14.36	13.43	12.84	11.27	14.02	13.89	11.73	12.67 1	12.97 1	12.20	13.76 14	.46 12.	74 13.90	13.8	2 13.92	2 11.51	11.26	12.69	11.69	10.96	10.15	11.48	11.81	13.45	10.34	13.65	180 7	23
Hu, col1030.36 <th0.36< th="">0.36<th0.36< th="">0.36<th0.3< th=""><th>CaO</th><td>23.02</td><td>23.07</td><td>23.46</td><td>22.92</td><td>22.38</td><td>21.50</td><td>22.20</td><td>2225 2</td><td>22.61 2</td><td>22.99</td><td>21.97</td><td>22.74 22</td><td>68 23.(</td><td>5 23.18</td><td>3 23.2</td><td>0 22.99</td><td>20.77</td><td>22.71</td><td>23.37</td><td>22.44</td><td>22.94</td><td>23.80</td><td>23.41</td><td>24.19</td><td>24.31</td><td>23.96</td><td>24.38 2</td><td>1.34 23</td><td>8</td></th0.3<></th0.36<></th0.36<>	CaO	23.02	23.07	23.46	22.92	22.38	21.50	22.20	2225 2	22.61 2	22.99	21.97	22.74 22	68 23.(5 23.18	3 23.2	0 22.99	20.77	22.71	23.37	22.44	22.94	23.80	23.41	24.19	24.31	23.96	24.38 2	1.34 23	8
K,C10100000000001101101001	Na ₂ O	0.31	0.29	0.36	0.35	0.43	0.33	0.36	0.64	0.24	0.29	0.45	0.35 G	.33 0.	t0 0.3	2	9 0.1	8 0.23	0.31	0.28	0.36	0.37	0.47	0.45	0.44	0.42	0.41	0.40	138 C	36
Math1011385010238481781781681781781781310078131007813100781310078131007813100781310078131007813100781310078131007813100781310078131007814100781410078141007814100781410078141007813101710131011 <th>K₂0</th> <td>0.01</td> <td>0.00</td> <td>0.00</td> <td>0.00</td> <td>0.02</td> <td>0.01</td> <td>0.01</td> <td>0.01</td> <td>0.00</td> <td>0.01</td> <td>0.00</td> <td>0.02</td> <td>.01 0.1</td> <td>0.0</td> <td>1 0.0</td> <td>0.0</td> <td>1 0.02</td> <td>0.01</td> <td>0.00</td> <td>0.02</td> <td>0.02</td> <td>0.00</td> <td>0.00</td> <td>0.01</td> <td>0.04</td> <td></td> <td>0.03</td> <td>002 C</td> <td>90</td>	K ₂ 0	0.01	0.00	0.00	0.00	0.02	0.01	0.01	0.01	0.00	0.01	0.00	0.02	.01 0.1	0.0	1 0.0	0.0	1 0.02	0.01	0.00	0.02	0.02	0.00	0.00	0.01	0.04		0.03	002 C	90
M188188173189183183183183183183183184173183184173184173184173183184173184173183 <th< th=""><th>Total</th><td>100.13</td><td>99.86</td><td>100.02</td><td>99.24</td><td>99.18</td><td><u> 98.68</u></td><td>99.17</td><td>39.07</td><td>39.18 5</td><td>39.56</td><td>38.83 10</td><td>0.22 95</td><td>.300 100.6</td><td>1 100.2</td><td>96.5</td><td>4 99.0</td><td>7 99.43</td><td>99.31</td><td>100.07</td><td>99.62</td><td>100.08</td><td>98.09</td><td>98.04</td><td>99.13</td><td>99.27</td><td>00.22</td><td>99.42 100</td><td>46 96</td><td>89</td></th<>	Total	100.13	99.86	100.02	99.24	99.18	<u> 98.68</u>	99.17	39.07	39.18 5	39.56	38.83 10	0.22 95	.300 100.6	1 100.2	96.5	4 99.0	7 99.43	99.31	100.07	99.62	100.08	98.09	98.04	99.13	99.27	00.22	99.42 100	46 96	89
SIV1 (36) </th <th></th> <td></td> <td>-</td> <td></td> <td></td> <td></td> <td></td>																										-				
AIV01490149025002800370032002800320028	si"	1.851	1.809	1.763	1.738	1.699	1.823	1.828	1.697	1.745 1	1.770	1.715	1.833 1.4	373 1.8.	26 1.819	1.85	7 1.843	3 1.778	1.645	1.725	1.808	1.825	1.550	1.638	1.646	1.774	1.525	1.880 1	679 1.	474
M ¹ 0000 0010 <th< th=""><th>AIV</th><td>0.149</td><td>0.191</td><td>0.237</td><td>0.262</td><td>0.301</td><td>0.177</td><td>0.172</td><td>0.302</td><td>0.255 C</td><td>).230 (</td><td>3.285 (</td><td>0.167 0.</td><td>127 0.1</td><td>74 0.18</td><td>0.14</td><td>3 0.15</td><td>7 0.222</td><td>0.355</td><td>0.275</td><td>0.192</td><td>0.175</td><td>0.450</td><td>0.362</td><td>0.354</td><td>0.207</td><td>0.475</td><td>0.120 0</td><td>321 0.4</td><td>526</td></th<>	AIV	0.149	0.191	0.237	0.262	0.301	0.177	0.172	0.302	0.255 C).230 (3.285 (0.167 0.	127 0.1	74 0.18	0.14	3 0.15	7 0.222	0.355	0.275	0.192	0.175	0.450	0.362	0.354	0.207	0.475	0.120 0	321 0.4	526
M10056004600560005000700070007000700070007000700070007000700070007000700070007000700070007000100070001	е Ф								0.002																	0.019				
Ti 034 014 016 005 007 017 017 016 0105 015 015 015 015 015 015 016 010 010 010 010 010 010 010 010 010	AI ⁴	0.050	0.045	0.043	0.056	0.032	0.014	0.007	00000	C007 C	0.038	0.016 (0.057 0.0	338 0.0	26 0.033	3 0.04	2 0.038	3 0.015	0.078	0.044	0.047	0:030	0.016	0.023	0.006	0.000	0.007	0.007 0	019 0.0	80
Cr00060009	F	0.034	0.041	0.063	0.071	0.107	0.061	0.058	0.125 (0.068 C	0.071 () 1001	0:050 0:0	0.0	38 0.05*	0.04	6 0.045	5 0.092	0.113	0.078	0.082	0.077	0.169	0.128	0.114	0.081	0.189	0.066 0	007 0.0	207
Fe ⁴ 0.047 0.064 0.087 0.084 0.087 0.084 0.087 0.084 0.081 0.112 0.123 0.012 0.136 0.141 0.123 0.012 0.136 0.141 0.123 0.014 0.136 0.141 0.123 0.024 0.135 0.012 0.136 0.141 0.123 0.026 0.161 0.125 0.026 0.136 0.141 0.124 0.025 0.124 0.125 0.136 0.141 0.141 0.125 0.164 0.155 0.141 0.155 0.141 0.155 0.141 0.155 0.156 0.156 0.156 0.156 0.156 0.141 0.156 0.141 0.155 0.141 0.155 0.141 0.155 0.156 0.156 0.156 0.156 0.156 0.141 0.156 0.141 0.156 0.156 0.156 0.156 0.156 0.156 0.156 0.156 0.156 0.156 0.156 0.156 0.156 0.156 0.156 <t< th=""><th>ت ت</th><td>0.006</td><td>0.009</td><td>0.007</td><td>0.006</td><td>0.001</td><td>0.003</td><td>0.005</td><td>00000</td><td>0.015 C</td><td>) 600.0</td><td>) 100.0</td><td>0.016 0.0</td><td>113 0.0</td><td>0.00</td><td>0.00</td><td>9 0.010</td><td>~</td><td>0.016</td><td>0.012</td><td></td><td></td><td></td><td></td><td>0.001</td><td>0.001</td><td>0.001</td><td></td><td></td><td></td></t<>	ت ت	0.006	0.009	0.007	0.006	0.001	0.003	0.005	00000	0.015 C) 600.0) 100.0	0.016 0.0	113 0.0	0.00	0.00	9 0.010	~	0.016	0.012					0.001	0.001	0.001			
For 1000 0.036 0.057 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.056 0.	ъ Ф	0.047	0.074	0.087	0.084	0.087	0.063	0.071	0.101 (0.044 C	0.064 () 114 (002010	13 0.0-	11 0.060	0.01	4 0.032	2 0.041	0.059	0.084	0.010	0.019	0.132	0.117	0.152	0.094	0.119	0.012 0	316 0.4	44
	Fe ^{&}	0.050	0.038	0.057	0.067	0.135	0.072	0.082	0.110 (0.105 C) 960.0	0.085 (0.0990 0.0	396 0.1	0.085	9 0.12	1 0.095	9 0.198	0.102	0.078	0.204	0.257	0.085	0.079	0.050	0.062	0.101	0.149 0	0.0 0.0	010
Mg ² 0813 0772 0781 0771 0584 0787 0776 0756 0757 0756 0757 0756 <	Mn &																						0.002		0.001	0.002	0.002	0.003	006 0.0	900
Ca ² 0 307 0 314 0 332 0 200 0 391 0 320 0 391 0 393 0 393 0 397 0 396 0 374 0 398 0 376 0 376 0 307 0 307 0 307 0 307 0 307 0 307 0 307 0 307 0 307 0 307 0 307 0 307 0 307 0 307 <th< th=""><th>Mg*</th><td>0.813</td><td>0.792</td><td>0.743</td><td>0.717</td><td>0.638</td><td>0.787</td><td>0.777</td><td>0.664 (</td><td>0.711 C</td><td>0.723 (</td><td>) 0690</td><td>0.758 0.</td><td>797 0.7</td><td>0.76</td><td>0.76</td><td>8 0.776</td><td>0.655</td><td>0.633</td><td>0.705</td><td>0.657</td><td>0.617</td><td>0.582</td><td>0.654</td><td>0.665</td><td>0.751</td><td>0.581</td><td>0.759 0</td><td>554 0.</td><td>415</td></th<>	Mg*	0.813	0.792	0.743	0.717	0.638	0.787	0.777	0.664 (0.711 C	0.723 () 0690	0.758 0.	797 0.7	0.76	0.76	8 0.776	0.655	0.633	0.705	0.657	0.617	0.582	0.654	0.665	0.751	0.581	0.759 0	554 0.	415
Na* 0 002 0 002 0 002 0 002 0 002 0 003 0	Ca ²⁺	0.907	0.914	0.932	0.920	0.910	0.868	0.892	0.904 (3.912 C).922) 833	0.1901 0.1	399 0.9	16 0.919	0.92	6 0.92	2 0.849	0.918	0.933	0.907	0.928	0.980	0.958	0.979	0.976	696.0	0.974 0	989 0.9	066
K* 0000 0001 0000	Nat	0.022	0.021	0.026	0.026	0.032	0.025	0.026	0.047 (0.018 C	0.021	0.033	0.025 0.0	724 0.0.	29 0.023	0.01	4 0.013	3 0.017	0.023	0.020	0.026	0.027	0.035	0.033	0.032	0.030	0:030	0.029 0	028 0.0	727
Mn ² 0.002 0.002 0.002 0.002 0.002 0.001 <th< th=""><th>, ₹</th><td>0.000</td><td></td><td></td><td>0.001</td><td>0.001</td><td>0.000</td><td>0.001</td><td>0.00</td><td>0</td><td>000.0</td><td></td><td>0.001 0.0</td><td>0.0</td><td>0.000</td><td>0</td><td>0.00</td><td>0.001</td><td>0.000</td><td>0.000</td><td>0.001</td><td>0.001</td><td></td><td></td><td>0.000</td><td>0.002</td><td></td><td>0.001 0</td><td>0.0</td><td>g</td></th<>	, ₹	0.000			0.001	0.001	0.000	0.001	0.00	0	000.0		0.001 0.0	0.0	0.000	0	0.00	0.001	0.000	0.000	0.001	0.001			0.000	0.002		0.001 0	0.0	g
Profile Role 0.005 0.004 0.005 0.004 0.005 0.004 0.005 0.004 0.005 0.004 0.005 0.006 0.006 0.005 0.006 0.006 0.005 0.006 <t< th=""><th>Mn^{2*}</th><td>0.002</td><td>0.002</td><td>0.002</td><td>0.002</td><td>0.004</td><td>0.004</td><td>0.004</td><td>0.005 (</td><td>0.003 C</td><td>0.003</td><td>0005</td><td>0.004 0.0</td><td>0.0</td><td>0:00</td><td>0.00</td><td>3 0.00</td><td>1 0.005</td><td>0.001</td><td>0.003</td><td>0.006</td><td>0.007</td><td></td><td>0.001</td><td></td><td></td><td>0.001</td><td></td><td></td><td></td></t<>	Mn ^{2*}	0.002	0.002	0.002	0.002	0.004	0.004	0.004	0.005 (0.003 C	0.003	0005	0.004 0.0	0.0	0:00	0.00	3 0.00	1 0.005	0.001	0.003	0.006	0.007		0.001			0.001			
Wo(k) 4807 4853 5009 4997 49.7 455 5009 4997 49.7 45.7 4889 4944 4948 4952 48.7 48.6 48.6 47.7 48.8 48.7 48.6 48.6 45.7 51.8 5052 49.14 49.78 55.05 22.75 53.01 51.22 54.70 51.30 50.90 53.10 50.1 24.0 14.0 14.0 14.0 14.0 14.0 14.0 14.0 1	Бе Ф	0.068	0.063	0.040	0.051	0.054	0.103	0.076	0.043 (0.067 C	0.064 (0.068	0.069 0.0	74 0.0	50 0.047	0.05	7 0.06	0.128	0.058	0.045	0.061	0.036		0.008						
En/%) 43.07 42.02 38.99 38.94 4.300 4150 40.82 36.28 38.60 38.66 37.19 40.96 42.36 37.56 40.69 40.64 41.00 34.91 35.77 38.19 35.64 33.09 32.68 35.96 35.09 35.04 35.64 33.09 32.68 35.96 35.04 30.01 28.50 23.00 Fe/%) 886 944 10.02 11.09 15.30 12.7 11.26 10.29 43.8 1122 1153 14.68 10.57 9.89 1351 10.52 10.34 10.32 19.82 12.40 1130 15.22 17.13 12.77 11.26 10.96 933 12.60 8.70 2060 24.60	Wo (%)	48.07	48.53	50.09	49.97	49.79	45.76	46.89	49.44	19.48 4	19.52 7	18.12	18.67 47	75 48.1	33 48.79	49.0	2 48.6	3 45.27	51.83	50.52	49.14	49.78	55.05	52.75	53.01	51.23	54.70	51.30 5	53 061	10
F5(%) 8 66 3 44 1002 1100 1530 274 1229 1428 1162 1163 1468 1037 9.88 1361 1052 10.34 1032 13.82 12.40 1130 1522 17.13 1227 11.26 1098 9.33 12.60 8.70 2060 2460	En (%)	43.07	42.02	39.90	38.94	34.90	41.50	40.82	36.28	38.60 3	38.86	37.19 4	t0.96 42	36 37.	56 40.65	40.6	4 41.00	34.91	35.77	38.19	35.64	33.09	32.68	36.99	36.01	39.45	32.80	40.01 2	3.50 22	30
	Fs (%)	8.86	9.44	10.02	11.09	15.30	12.74	12.29	14.28	11.92 1	11.63	14.69	10.37 5	.89 13.	31 10.52	2 10.3	4 10.32	2 19.82	12.40	11.30	15.22	17.13	12.27	11.26	10.98	9.33	12.60	8.70 2	0.60 24	8

TABLE 2

Representative clinopyroxene microprobe analyses of Sal lavas

Ged Unit		Main	Eruptive F	ormation				SerraN	egra						Monte	arande - P	edra Lume					
Lithotype	Nephelinite	a,					Basalt				Nephelinite	į	Veph. 1	Basalt			ephelinite	N	ephelinite			
Sample	116						109				238	4	124 (582		3	67	1	315			
Type	Crsp	Crsp	Magn	M agn	Magn	M agn	Chr	Chr	Chr	Magn	Crsp	M agn	Crsp	Magn	M agn	M agn	M agn	M agn	Magn	Magn	Magn	Magn
	입	υ	₽2	_	Σ	рс	Σ	입	ů	р Ц	ů	Σ	_	Σ	Σ	Σ	Σ	Σ	Σ	Σ	hc	Σ
AI ₂ O ₃ w%	40.68	44.14	13.80	523	256	295	9.65	15.62	17.20	6.19	45.51	8.20	56.88	3.80	2.81	483	5.68	4.76	3.82	1.69	3.89 3.89	<u>18</u>
	2.10	0.33	7.66	15.57	18.25	17.41	7.43	3.16	3.12	16.82	1.27	17.77	0.72	26.05	26.75	25.47	14.53	16.07	18.38	20.37	18.11	20.43
رت م	13.42	19.92	15.70	4.62	1.85	223	24.01	31.21	32.08	604	12.86	4.47	1.07	0.0 0	000	000	3.35	1.06	1.55	90:0	1.40	0.07
د _ء 0 م	0.11	60.0	0.26	0.35	0.40	0.42	0.25	0.17	0.18	051	0.16	0.54	0.06	0.45	020	0.74	0.32	0.31	0.42	0.43	620	0.45
Fe ₂ 03	11.11	5.56	25.40	30.06	31.41	31.60	21.11	15.80	15.32	23.93	7.59	21.67	10.28	13.35	13.95	14.53	31.52	32.68	29.92	29.41	30.37	29.31
FeO	16.22	10.92	27.04	35.54	38.60	37.81	30.05	24.22	19.38	41.00	16.86	41.70	11.98	53.56	53.66	51.52	37.56	37.87	<u>39.30</u>	41.54	37.97	41.13
Onin	0.17	0.39	020	0.58	0.71	0.61	0.38	0.00	0.25	020	0.29	0.73	0.18	0.67	0.75	0.64	0.69	0.76	0.89	1.10	081	100
OgM	15.40	18.15	8.37	6.57	589	5.86	6.00	7.92	11.36	392	15.01	4.16	19.29	1. 14	1.14	221	4.37	5.04	5.58	484	6.18	523
Total	99.21	<u> 99.50</u>	98.74	99.13	69:66	<u> 06:96</u>	<u>98.89</u>	98.40	<u> 06:90</u>	98.91	<u> 99:55</u>	99.24	100.46	99.94	<u> 99.26</u>	<u> 36:95</u>	<u> 98.03</u>	<u> 98.5</u> 6	<u> 98'66</u>	99.44	99.11	99.43
AI+3	10.933	11.475	4.371	1.744	0.864	1.002	3.158	4.921	5.232	2:092	11.995	2729	13.974	1.304	0.978	1.641	1.943	1.620	1.281	0.578	1.307	0.624
T +4	0.360	0.055	1.549	3.310	3.929	3.772	1.562	0.635	0.606	3.628	0213	3.773	0.113	5.833	5.938	5.519	3.168	3.487	3.933	4.443	3.885	4,441
ۍ ئ	2.420	3.475	3.336	1.033	0.418	0507	5271	6.594	6.544	1.370	2275	0.997	0.176	0.001	0000	0000	0.769	0.241	0.350	0.015	0.315	0.015
۷+3	0.021	0.015	0.057	620:0	0.093	8600	0.057	0.037	0.037	0.118	0.028	0.123	0.010	0.106	0.047	0.171	0.073	0.072	0:096	660.0	0.088	0.103
Fe ⁺³	1.906	0.923	5.138	6.523	6.766	6850	4,411	3.177	2.975	5.163	1277	4.605	1.613	2.924	3.099	3.150	6.878	7.094	6.407	6.422	6.520	6.375
Fe +2	3.092	2015	6.079	8.404	9.240	9.108	6.978	5.412	4.182	9.832	3.153	9.847	2088	13.037	13.249	12.412	9.110	9.135	9.352	080.01	090.6	9.942
Mn ⁺²	0.033	0.072	0.115	0.139	0.173	0.148	0600	690:0	0.054	0.122	0.055	0.175	0.032	0.164	0.186	0.156	0.170	0.185	0.215	0.269	0.195	0.244
Mg ⁺²	5.235	5.968	3.356	2.768	2.515	2515	2,484	3.154	4.370	1.674	5:005	1.751	5.993	0.631	0.503	0.950	1.889	2167	2.366	2.095	2.630	2.255
molUsp	0.262	0.106	0.376	0.538	0.464	0.453	0.413	0.286	0.290	0.584	0250	0.621	0.123	0.798	0.791	0.775	0.452	0.441	0.490	0.517	0.471	0.514
C – core. R	- rim. I -	- interme	diate zon	le. M – r	natrix. Ir	1c - inclu	o ui noist	livine. C	r-sp – ch	romian	spinel: C	hr – chro	mites: N	Лаеп — Т	i – magn	etites. Fc	r the com	nolete se	t of analy	'ses see	ORRES (1998).
,									4					b	2							

 TABLE 3

 Representative oxide microprobe analyses of Sal lavas

Geol Unit	Main	Eruptive	Ŀ		Sen	ra Negra					ſ	^o onta do	A tar Ba	leia				2	onte Gr	ande Pe	dra - Lur	e	Γ
Sample	115			109			1025		619				É	60			ø	R					
	ပ	Σ	Σ	Σ	Σ	Σ	Σ	Σ	ပ	F	2	2	Σ	ပ		к	Σ	Σ	Σ	Σ	Σ	Σ	Σ
SiO ₂ wt%	53.39	57.73	51.30	52.97	51.85	51.70	61.04	51.57	50.07	49.90	52 88	49.01	54.98	49.60	49.73	51.23	51.25	49.04	49.09	52 06	48.81	48.81	49.47
A₂0₃	29.43	23.06	20 48	28.52	<u>28</u> 28	29.46	23.42	29.57	<u>30</u> 80	31.07	<u>8</u> 0 20	29.82	27.74	31.22	30.92	<u>30</u> .03	30.05	31.56	31.23	P2 62	31.66	31.62	31.03
TIO ₂	0.19	0.15	0.17	0.21	0.20	0.19	0.16	0.21	0.11	0.07	0.12	0.13	0.15	0.13	0.12	0.17	0.14	0.13	0.12	0. 15	0.13	0.13	0.15
Fe ₂ O ₃	0.97	0.60	0.87	0.75	1.00	0.90	0.34	0.82	0.48	0.58	0.41	0.60	0.42	0.57	0.64	0.65	0.73	0.59	0.59	0.47	0.52	0.53	0.62
MgO	0.09 0	0.08	0 30	0.0	0.13	0.09	0.32	0.04	0.08	0.07	0.00	0.06	0.00	0.13	0.24	<u>0</u> 13	0.11	0. 12	0.08	0 0	0.08	<u>0</u> 00	0.0
MnO	8 0	0.0	0.0	0.02	0.02	0.03	0 0	0.01	0 0	00.0	0 0	000	0 0	0.01	0 0	0 0	ОO	0.05	0.01	0.04	0.03	<u>0</u> 0	0.02
BaO	00 0	22	<u>0</u> 0	0.11	<u>0</u> 0	0.11	0.62	0.00	0.01	0.19	0.13	0.02	0.15	0.08	0.13	0 0	0.16	0.05	0.14	0.08	0.10	<u>0</u> 0	0.07
CaO	11.90	4, 13	11.67	11.44	12.54	12.60	5.06	12.64	13.89	14.32	11.28	13.73	9.78	14.45	14.33	12.87	13.49	14.71	15.02	12.42	15.28	15.33	14.74
Na ₂ O	3.75	3.61	3.81	4.37	3.86 3.86	3.79	<u>6</u> 90	3.73	3.18	281	4 4	3.11	5.30	281	282	3.50	3.50	271	2.76	3.86 3	2.62	261	2.93
K₂O	0.37	7.51	0.58	0.48	0.49	0.37	1.83	0.26	0.17	0.17	0.34	0.24	0.52	0.23	0.22	0.31	0.33	0.23	0.19	0.38	0.24	0.23	0.26
Total	100.15	<u>80</u> .08	<u>86</u> 30	<u> 36</u> .96	<u> 98</u> .73	99.25	<u>80</u> 00	<u>98.95</u>	<u>98</u> 80	99.17	98.74	96.72	<u>80</u> 00	99.24	<u> 39</u> .14	<u> 99.</u> 02	66.77	<u> 90</u> 19	<u>99.</u> 23	<u> 90</u> 10	99.47	<u> 36</u> 36	<u>86</u> .38
-2 -2	100 0				0		100	00		000			5					000		00			
ca		C78.0	2311	147.7	7480	2.4//	U.9/4	Z 480	5	7.873		Q//7	1.50/		7.873	7.037	707	29062	7.8/7	2430	3.01/	3.028	7 210
Si [‡]	9.656	10.760	9.483	9.713	9.570	9.485	10.967	9.478	9.235	9.186	9,604	9.249	10.013	9.131	9.165	9.410	9.374	9.045	9.064	9.532	8,996	<u>9,000</u>	9.113
A ³	6.273	5.064	6.423	6.163	6.216	6.372	4,960	6.405	6.005	6.740	6.285	6.632	5.964	6.773	6.715	6.501	6.477	6.859	6.795	6.409	6.877	6.871	6.738
¥.	0.026	0.020	0.023	0.029	0.027	0.026	0.021	0.030	0.015	0.010	0.016	0.018	0,020	0.017	0.016	0.023	0.019	0.018	0.017	0.020	0.019	0.018	0.021
Fe.3	0.130	0.084	0.120	0.104	0.138	0.124	0.046	0.113	0.067	0.080	0.057	0.085	0.057	0.079	0.088	0:060	0,100	0.081	0.082	0.064	0.072	0.073	0.085
Mg ⁺²	0.017	0.185	0. 107	0.025	0.037	0.026	0.087	0.012	0.023	0.020	0.016	0.017	0.004	0.036	0.066	0.036	0.031	0.034	0.022	0.010	0.021	0.025	0.025
Mn ^{±2}	000	0000	0.003	0.003	0.003	0.005	0.001	0.002	000.0	000.0	000	0.000	0,000	0.002	000.0	0.000	0.001	0.008	0.001	0.006	0,004	0.001	0.003
Ba⁺²	0.007	0. 162	0.00	0.008	0.005	0.008	0.044	0.006	0.001	0.013	0000	0.002	0.010	0.006	0000	0.002	0.012	0.004	0.010	0.006	0.007	0.001	0.005
Ca⁺²	2.305	0.825	2311	2247	2480	2.477	0.974	2.490	2744	2.823	2215	2.775	1.907	2.851	2 829	2.532	2644	2906	2972	2436	3.017	3.028	2910
Na⁺¹	1.316	1.305	1.306	1.554	1.381	1.350	2402	1.330	1.138	1.004	1.577	1.139	1.872	1.004	1.007	1.278	1.241	0.968	0.988	1.371	0.938	0.934	1.047
₽ţ	0.086	1.787	0.136	0.113	0.114	0.086	0.420	0.062	0.041	0.040	0.080	0.058	0.120	0.054	0.062	0.073	0.076	0.054	0.045	0.089	0.056	0.0555	0.061
% An	60.24	23.68 23	61,70	57,60	60 60	63 46	27 CB	64 17	01 07	06 62	57.24	69.97	48 84 14	73 10	73.04	65 48	88	74 18	74 17	88	75 23	75.52 75	70.52
% Ab	35.27	30.61	34.81	8	94 36	34.16	61.15	34,08	88	25.75	40.46	28.54	47.83	25.40	25.41	32 59	80.00	24.36	24.46	35.00	23.20	23.10	25.84
% Or	249	45.71	3.49	3.06	295	2.38	11.82	1.75	1.07	1.36	230	1.49	3.33	1.51	1.55	1.93	2 19	1.46	1.37	242	1.57	1.38	1.64
C – core, F	t – rim, I	- interr	nediate	zone, M	I – micro	olite in g	groundm	lass. Fo.	T the co	nplete st	et of ane	ulyses se	e Torre	s (1998									

 TABLE 4

 Representative plagioclase microprobe analyses of Sal Island basalts

Geol Unit	Main E	ruptive	Form.	P. Altar	Baleia	M.GrP	. Lume	Geol Unit	P. Altar	Baleia	I	M.Gran	de - Pedr	a Lume	
Sample	116			424		1315		Sample	424		1315		1248		
SiO ₂ (wt%)	41.37	40.89	40.04	41.08	41.14	40.89	41.34	SiO ₂ (wt %)	41.89	42.71	41.98	42.01	42.84	42.56	42.94
TiO₂	0.06	0.08	0.08	0.09	0.12	0.11	0.10	TiO ₂	0.19	0.08	0.06	0.12	0.08	0.07	0.06
Al ₂ O ₃	33.10	33.08	32.88	33.00	32.65	32.49	32.88	Al ₂ O ₃	6.82	6.65	6.31	7.15	6.79	6.79	6.56
Fe ₂ O ₃	0.84	1.03	1.02	1.07	0.92	1.01	1.01	FeO	3.11	3.35	3.27	3.51	3.11	3.07	3.47
MnO					0.01	0.03	0.02	MnO	0.15	0.06	0.09	0.04	0.05	0.08	0.11
MgO	0.17	0.16	0.16	0.15	0.19	0.15	0.17	MgO	8.13	7.58	7.76	7.29	8.01	8.05	7.46
CaO	0.77	0.81	2.82	0.34	0.33	0.47	0.50	CaO	34.26	33.81	34.28	33.29	34.50	34.38	34.18
Na₂O	15.19	14.35	14.16	13.76	13.36	14.31	14.14	Na₂O	2.84	3.17	2.90	3.19	3.03	3.14	3.32
K₂O	7.57	7.72	7.25	9.84	9.83	8.51	8.31	K₂O	0.15	0.16	0.14	0.12	0.08	0.10	0.08
BaO	0.21	0.42	0.38	0.67	0.52	0.75	0.49	BaO	0.13	0.60	0.52	0.55			
Total	99.27	98.52	98.79	99.99	99.07	98.72	98.95	Total	97.67	98.17	97.32	97.27	98.49	98.24	98.18
Si ⁺⁴	8.122	8.095	7.946	8.099	8.162	8.135	8.161	Si⁴⁺	3.895	3.958	3.933	3.928	3.937	3.924	3.965
Ti ⁺⁴	0.009	0.012	0.012	0.013	0.018	0.016	0.015	Ti⁴⁺	0.013	0.006	0.004	0.008	0.006	0.005	0.004
AI ⁺³	7.659	7.720	7.691	7.669	7.634	7.616	7.650	Al ³⁺	0.747	0.726	0.697	0.788	0.735	0.738	0.714
Fe ⁺³	0.123	0.153	0.153	0.158	0.138	0.151	0.150	Fe ²⁺	0.241	0.260	0.256	0.275	0.239	0.237	0.268
Mn ⁺²					0.001	0.006	0.003	Mn ²⁺	0.012	0.005	0.007	0.003	0.004	0.006	0.009
Mg ⁺²	0.050	0.046	0.048	0.043	0.058	0.043	0.050	Mg ²⁺	1.127	1.047	1.084	1.016	1.097	1.107	1.027
Ca ⁺²	0.161	0.171	0.600	0.072	0.070	0.100	0.106	Ca ²⁺	3.413	3.357	3.441	3.335	3.397	3.396	3.381
Na ⁺¹	5.783	5.507	5.450	5.261	5.139	5.519	5.412	Na⁺	0.513	0.569	0.526	0.578	0.540	0.561	0.594
K ⁺¹	1.895	1.949	1.835	2.474	2.488	2.160	2.092	K⁺	0.017	0.019	0.017	0.014	0.009	0.012	0.009
Ba ⁺²	0.016	0.033	0.029	0.052	0.041	0.059	0.038	Ba ²⁺	0.005	0.022	0.019	0.020			
% Ne	72.8	70.6	72.4	64.9	63.6	68.7	68.3	% Aker	66.5	66.0	67.5	63.9	66.3	66.3	66.0
% Ks	26.6	27.8	27.6	34.0	34.3	29.9	29.4	% Gehl	33.5	34.0	32.5	36.1	33.7	33.7	34.0
% Qz	0.6	1.6	0.0	1.1	2.1	1.3	2.4								

 TABLE 5

 Representative nepheline and melilite microprobe analyses of Sal nephelinites



Fig. 4 - NiO and CaO (wt %) vs. Fo diagrams for olivine in Sal basaltic (filled squares) and nephelinitic (open squares) lavas.

nephelinites (Ca > 0.94 a.f.u.) pointing out to a greater importance of non-quadrilateral substitutions in nephelinites.

Pyroxene chemistry indicates that the S4 type $(CaMg_{0.5}Ti_{0.5}AI^{IV}SiO_6)$ is the most important nonquadrilateral substitution in the Sal pyroxenes (see Fig. 5). Sub-solidus esseneites, given their low Ti and Cr, and high Fe³⁺ relatively to Al in M1 position, are dominated by FATS (Fe³⁺-Al Tschermark or esseneite) component substitution (Fig. 5).

Oxides are predominantly members of the magnetite-ulvospinel solid solution series. Ti-magnetites



Fig. 5 - Histograms of relative percentages of the various types of substitution in pyroxenes (MORIMOTO et al. 1988) from basaltic and nephelinitic rocks.

Geol Unit	Main E. F.	Ponta c	lo Altar Ba	leia	Mte. Gra	nde - Pedra	a Lume
Lithotype	Neph	Nephelinite			Nephelinit	е	
Sample	116	424			1315	1247	1248
TiO₂ (wt %)	55.49	55.14	55.55	56.03	56.07	55.24	55.91
Al ₂ O ₃	0.17	0.67	0.48	0.21	0.26	0.23	0.20
Cr₂O₃	0.08			0.05			
FeO	0.94	1.36	1.62	1.06	1.04	1.35	1.06
MnO	0.01		0.01		0.05	0.08	0.01
MgO	0.04	0.16	0.10	0.04	0.03	0.03	0.04
CaO	37.83	38.36	39.19	38.73	37.81	39.33	38.25
Na₂O	0.44	0.44	0.22	0.20	0.36	0.35	0.24
K₂O	0.02	0.08	0.06	0.01	0.04	0.01	0.01
Total	95.02	96.21	97.22	96.32	95.65	96.62	95.72
Ti ⁺⁴	7.9691	7.8320	7.8241	7.9411	7.9916	7.837	7.969
AI ⁺³	0.0382	0.1490	0.1065	0.0476	0.0576	0.051	0.045
Cr ⁺³	0.0114			0.0079			
Fe ⁺²	0.1507	0.2150	0.2534	0.1663	0.1646	0.213	0.168
Mn ⁺²	0.0021		0.0010		0.0082	0.013	0.002
Mg ⁺²	0.0118	0.0450	0.0282	0.0106	0.0096	0.008	0.011
Ca ⁺²	7.7396	7.7630	7.8631	7.8202	7.6779	7.95	7.768
Na ⁺¹	0.1611	0.1600	0.0783	0.0716	0.1318	0.128	0.088
K ⁺¹	0.0051	0.0200	0.0144	0.0032	0.0085	0.002	0.002

 TABLE 6

 Representative perowskite microprobe analyses of Sal lavas

TABLE 7	Major and trace element analyses of Sal Islan
TABLE 7	INIAJOT AND ITACE CLEMENT ANALYSES OF SAL ISLAND

lavas

	Main	Eruptive	: Formati	on	Po	inta Altar	· Baleia C	omplex		Se	rra Negr	a		Monte	Grande.	- Pedra L	ume	
Sample	115	475	116	103	1465	1001	6/9	676	424	13	1025	109	682	197	1315	238	367	192
косктуре	basalt	Dasait	m_nepn_m	wepn_m	basalt	Dasait	basait	Nepn_m	Neiline	basalt	basalt	Dasait	Dasalt		Melline	INI_nepn	IN_nepn	IN_nepn
SiO 2 (wt%)	42.89	39.46	39.64	38.43	45.24	41.54	39.90	38.97	34.17	45.71	4521	42.39	43.25	36.83	34.85	39.78	39.08	37.03
TIO ₂	3.62	4.48	2.78	3.85	2.56	2.79	2.95	3.43	4.08	3.17	3.40	3.73	2.81	3.36	3.74	3.89	3.36	3.91
AI 203	16.05	13.75	11.55	10.59	13.57	13.03	13.96	9.92	10.63	12.97	12.94	12.63	12.96	11.19	10.89	12.09	11.33	11.02
Fe ₂ O ₃	5.37	6.57	4.66	5.24	2.20	4.50	8.12	4.71	631	4.75	4.70	4.70	3.28	5.72	6.34	4.91	529	831
FeO	5.99	7.34	7.83	7.53	9.79	8.17	529	7.46	592	7.10	6.33	7.66	8.75	6.24	5.92	6.69	5.92	3.52
MnO	0.23	0.22	0.18	0.20	0.18	0.18	025	020	023	0.17	0.18	0.18	0.19	0.21	0.22	0.21	0.18	0.19
NgO	6.45	7.51	12.74	14.49	10.61	12.16	9.85	16.40	1325	10.70	10.50	10.81	12.08	12.97	13.01	12.25	14.61	14.66
CaO	12.90	12.42	12.54	14.41	11.50	11.33	12.73	13.48	1627	10.57	11.48	11.66	11.95	16.08	15.98	13.12	12.79	12.07
Na ₂ O	1.98	2.70	4.28	2.37	2.07	2.47	1.65	2.50	3.50	1.93	1.84	2.54	1.98	2.64	3.54	3.41	2.88	2.78
K20	1.17	0.99	1.39	0.74	0.83	0.97	0.89	129	0.98	0.75	122	1.19	1.30	0.48	1.18	1.23	0.78	0.59
P_05	0.66	0.54	0.76	0.86	0.37	0.45	0.31	0.82	1.69	0.35	0.49	0.48	0.44	1.68	1.65	1.01	0.93	0.72
H ₂ O							3.55											
LOI		4.00	1.66	1.30	1.08	2.41	080	0.82	298	1.83	1.72	2.03	1.00	2.60	2.66	1.42	2.86	520
Total	97.31	<u> 96'66</u>	100.01	100.01	100.01	100.00	100.25	100.00	100.01	100.00	100.01	100.00	66.66	100.00	96.66	100.01	100.01	100.00
# BW	51.51	5025	65.38	67.83	61.63	63.94	58.22	71.41	67.13	62.64	63.93	61.84	64.78	66.99	66.60	66.28	70.91	70.37
(U J	10	OCK	107	7077	100	Cat.	GEO	146	006	200	010	376	246	760	00	100	360
	3 5	5 6	440	101	<u> </u>		100		0 <u>+</u>			200	300	0 7	107	100	- 00C	800 620
z	9/	70	700	0000 0000	3	88	<u>8</u>	0000	cuc	787	3	R	607	117	117	007	007	C/7
Sc	8	19	3	24	15	26	21	25	8	24	25	24	26	33	25	24	24	27
Cu	122	28	83	133	118	49	163	127	95	127	104	54	117	129	02	100	6/	49
Zn	110	83 83	95	96	103	97	66	96	101	128	100	6	86	66	88	101	86	74
Rb	27	21	18	24	13	17	27	35	18	16	27	18	27	8	34	15	9	20
Cs	0.21	0.42	0.41	0.20	0.20	0.20	021	020	021	020	0.20	0.20	0.20	0.21	0.41	0.61	0.51	0.53
Ba	423	412	583	689	158	340	389	559	1082	278	385	428	341	752	1002	599	749	598
Sr	954	694	1010	1099	521	468	624	1030	1898	511	593	535	596	1497	1457	991	896	676
Та	2.77	4.06	3.15	3.14	1.62	2.66	3.02	4.03	4.74	1.94	2.85	2.86	2.73	4.00	5.03	4.26	3.19	4.11
qN	39.1	57.3	59.0	45.6	31.3	42.0	37.5	51.4	96.9	275	32.6	42.9	36.4	61.6	85.3	54.8	60.7	59.1
Ŧ	4.32	4.69	4.47	5.27	2.53	4.51	521	534	587	428	5.29	5.31	4.65	5.03	6.37	5.78	494	5.80
Zr	224	196	236	250	189	181	248	250	376	199	237	218	224	321	329	305	270	265
<u>×</u>	26	25	24	24	18	17	24	24	46	19	33	3	8	29	37	23	30	23
ЧL	2.90	2.90	5.10	5.50	1.00	1.80	2.00	5.00	10.10	1.90	2.10	1.90	1.80	8.00	10.00	4.40	5.40	3.30
D	0.82	0.83	1.32	1.32	0.40	0.82	0.73	121	3.09	0.41	0.41	0.71	0.71	2.05	277	122	134	1.48
La	39.3	37.7	57.4	57.3	13.8	24.1	28.3	59.3	116.0	21.1	25.5	24.9	25.5	89.8	109.0	51.5	63.5	40.6
ő	62	17	102	103	29	50	09	115	201	44	54	54	52	157	197	94	116	79
PN	37	36	43	47	15	28	30	51	62	53	31	28	26	65	11	44	51	37
Sm	7.32	7.32	8.24	8.58	3.14	5.66	6.55	9.77	14.60	5.36	6.63	6.54	5.74	12.40	14.50	8.67	939	7.78
Εu	2.25	2.37	2.73	2.97	1.20	2.00	226	325	4.60	1.84	2.28	2.12	1.98	3.83	4.57	2.79	3.00	259
ЧP	1.00	1.00	1.00	120	09:0	06.0	1.00	120	1.70	080	0:00	1.00	0.80	1.50	1.70	1.20	1.10	1.00
ЧÞ	2.03	1.90	1.46	1.47	1.12	1.75	222	1.66	229	1.56	1.76	1.68	1.66	2.01	2.53	1.87	1.7.1	1.70
Lu	0.26	0.26	0.18	0.20	0.14	0.22	027	021	029	0.19	0.22	0.21	0.21	0.27	0:30	0.25	023	022

P. Torres; L. C. Silva; J. Munhá; R. Caldeira; J. Mata & C. Tassinari

Neph_m - nephelinite with melilite; M_neph - melanephelinite. For the complete set of analyses see TORRES (1998).

Geol. Unit	Sample	Rock type	⁸⁷ Sr/ ⁸⁶ Sr	¹⁴³ Nd/ ¹⁴⁴ Nd	۶Nd
	115	Basalt	0.703350 ± 10	0.513029 ± 22	7.63
MEE	475	Basalt	0.703460 ± 10		
	116	Neph mel	0.703200 ± 10	0.512802 ± 33	3.20
	103	Neph mel	0.703080 ± 8	0.512888 ± 31	4.88
	1465	Basalt	0.703220 ± 8		
	1001	Basalt	0.703380 ± 10	0.512774 ± 26	2.65
PAB	679	Basalt	0.703370 ± 6	0.512771 ± 30	2.59
	676	Neph mel	0.703230 ± 7		
	424	Melilitite	0.703390 ± 14	0.512927 ± 25	5.64
	13	Basalt	0.703390 ± 7		
SN	1025	Basalt	0.703380 ± 6	0.512770 ± 56	2.57
	109	Basalt	0.703380 ± 6	0.513000 ± 24	7.06
	682	Basalt	0.703210 ± 10	0.512885 ± 29	4.82
	197	Melilitite	0.703230 ± 9	0.513027 ± 34	7.59
MGDI	1315	Melilitite	0.703130 ± 10	0.512900 ± 24	5.11
WIGFL	238	Melaneph	0.703670 ± 9	0.512971 ± 22	6.50
	367	Melaneph	0.703580 ± 7	0.512981 ± 26	6.69
	192	Melaneph	0.704101 ± 10	0.513118 ± 25	9.36

 TABLE 8

 Sr and Nd isotopic analyses of Sal Island lavas

MEF – Main Eruptive Formation, PAB– Ponta do Altar Baleia Complex SN – Serra Negra Formation, MGPL – Monte Grande – Pedra Lume.

(TiO₂ = 14.53 – 26.75 wt %) display progressive substitution of Cr + Al (0.59 – 7.71 a.f.u.) by Ti, Fe³⁺ and Fe²⁺ (% mol Usp = 43 – 80) with the advance of magmatic evolution. A few Al-rich chromian spinels and chromites (in particular those included in olivine phenocryst cores) can be considered as earlier crystallization phases that fractionated under relatively high-pressure conditions (e.g. MATA & MUNHÁ, 2004). Their composition ranges within the spinel-magnesiochromite solid solution due to Cr \Leftrightarrow Al exchange.

Plagioclase phenocrysts range from bytownite (An_{73}) cores to labradorite (An_{57}) rims. Groundmass microlites are highly variable extending from compositions similar to those characterizing phenocrysts down to 24 mol % An, reflecting progressive anortite $(CaAl_2Si_2O_8)$ replacement by albite $(NaAlSi_3O_8)$ along the fractionation processes.

Nepheline contains up to 9.8 wt % K_2O corresponding to 34 mol % Ks.

Melilite ranges from 64 to 67 mol % of akermanite.

Perowskite quantitative analyses show considerable substitutions of Ca by rare earth elements (knopite) that

were analyzed only qualitatively. Moreover, there is evidence for the Nb and small amounts of Ta replacing Ti (dysanalyte and loparite).

6 – GEOCHEMISTRY

6.1 Major and trace-element results

The studied Sal volcanic rocks (see Table 7 for major and trace-element analysis results) have low silica contents (SiO₂ \approx 34 to \approx 46 wt %) and high alkalis concentrations (Na₂O up to 4.73%; K₂O up to 2.6%), plotting within the alkalic field on the TAS (total alkalis vs. SiO₂) diagram – Fig. 6 (LE BAS *et al.*, 1986, LE MAITRE, 2002). Na₂O/K₂O (wt %) in excess of 1 ($\overline{x} = 2.36 \% \pm 0.09$), allows to consider these lavas as sodic following the proposal of Middlemost (1975).

CIPW normative compositions range from *Ne* normative (1 - 20 %) to *Hy* normative (3 - 17 %), with the great majority of samples (88%) being silica undersaturated and with a considerable number (45%) of them corresponding to foidites (Fig. 6 and 7).



Fig. 6 – Silica – Alcalis (TAS) diagram (LE BAS *et al.*, 1986). Lines 1 and 2 according to IRVINE AND BARAGAR (1971) and KUNO (1966). Open symbols – nephelinites and melilitites. Filled symbols – basalts and basanites. Squares – Main Eruptive Formation, circles – Serra Negra Eruptive Formation, triangles – Ponta do Altar-Baleia Complex, diamonds (melilities and melilite nephelinites) and asterisks (nephelinites) from Monte Grande Pedra Lume Formation.



Fig. 7 – Discrimination diagram for melilititic, nefelinitic and basanitic rocks, according to LE BAS (1989). Symbols as in previous figure.

All the hyperstene normative samples (12 % of the analysed samples) belong to the Serra Negra Eruptive Formation and are also characterized by distinct trace element ratios (e.g. \bar{x} Y/Nb = 0.60 ± 0.02; \bar{x} (La/Sm)n = 2.49 ± 0.03; \bar{x} (La/Yb)n = 10.24 ± 0.39) when compared to those ratios of undersaturated lavas from younger and older formations \bar{x} Y/Nb = 0.49 ± 0.08; \bar{x} (La/Sm)n = 3.79 ± 0.71; \bar{x} (La/Yb)n = 20.65 ± 8.58). The majority of Serra Negra lavas plot in the basalt field of the TAS diagram (Fig. 6) occupying a position close to the KUNO (1966) and IRVINE AND BARAGAR (1971) compositional dividers between alkaline and tholeiitic suites, but clearly inside the alkali basalt field defined by BELLIENI *et al.* (1983; in: LE MAITRE *et al.*, 2002).

The studied lavas are strongly enriched in incompatible elements with La being 50 to 500 times more concentrated than in chondrites. As seen in Fig. 8, REE are significantly fractionated, with (La/Yb)n ranging from 8.84 to 36.34. This fractionation is inversely correlated with the degree of silica-undersaturation or silica content, increasing from alkali basalts to melilitites through basanites and nephelinites (Fig. 9).

Other incompatible trace elements concentrations also vary within a significant range and show an increase with both differentiation and degree of silica undersaturation. Sal mafic lavas present extended trace element patterns with the relative degree of enrichment for the different elements tendentiously reflecting the order of incompatibility usually determined for magma genesis processes of non-orogenic oceanic lavas. This is demonstrated, for example, by values of Zr/Y (4.19 to 13.72: $\overline{x} = 10.21 \pm 0.22$), Nb/Zr (0.14 to 0.60; $\overline{x} = 0.21 \pm 0.01$), and Ba/Sr (0.30 to 1.55; $\overline{x} = 0.79 \pm 0.04$) higher those reported for chondrites (McDonough & Sun, 1995). However, a significant difference arises from the patterns depicted by melilitic/nephelinitic rocks as compared with those of basalts (Fig.10). Contrasting with basalts, strongly undersaturated rocks are characterized by marked Rb, K, and Hf negative anomalies.

6.2 Sr and Nd Isotopic results

The obtained results (see Table 8) are somewhat variable, covering the following ranges: ${}^{87}\text{Sr}/{}^{86}\text{Sr} =$



Fig. 8 - Chondrite normalized (MCDONOUGH AND SUN, 1995) Rare Earth Elements pattern for studied lavas from Sal Island.



Fig. 9 – La/Yb vs. SiO₂ diagram. Open symbols – Basanites, nephelinites and melilitites. Filled symbols – Basalts. Squares – Main Eruptive Formation, circles – Serra Negra Eruptive Formation, triangles – Ponta do Altar-Baleia Complex, diamonds (melilities and melilite nephelinites) and asterisks (nephelinites) from Monte Grande Pedra Lume Formation.



Fig. 10 – Primitive mantle-normalized (SuN & McDoNOUGH, 1989) incompatible element diagrams for basaltic and melilititic/nephelinitic rocks. Symbols as in previous figure.

= 0.703080 - 0.704101 and ¹⁴³Nd/¹⁴⁴Nd = 0.512770 - 0.513118. Sr isotopic ratios tend to be negatively correlated with ¹⁴³Nd/¹⁴⁴Nd, with the majority of samples plotting in the depleted quadrant (ε Sr < 0; ε Nd > 0) of the Sr-Nd diagram (Fig. 11). This indicates the derivation from source(s) with time-integrated Rb and Nd depletion relative to Sr and Nd, respectively. Sal lavas tend to plot below the mantle array and close to the composition of the HIMU mantle component. However, some of the analysed samples are characterized by high ⁸⁷Sr/⁸⁶Sr, for a given ¹⁴³Nd/¹⁴⁴Nd, as compared with samples from other islands of the Cape Verde archipelago. This is interpreted as reflecting the effects of alteration, which modified the magmatic Sr isotopic signatures but not the Nd ones.

It is interesting to note that some of the samples (thought unaffected by alteration processes), present ¹⁴³Nd/¹⁴⁴Nd ratios down to 0.51277, being clearly lower than those typically assigned to the HIMU and DMM mantle source components, usually considered as the main contributors for the mantle sources of the Northern Cape Verde islands.

7 – DISCUSSION

7.1 Age constraints

The Sal geochronological data (TORRES *et al.*, 2002a; HOLM *et al.*, 2008; see also this paper: 3 – Geology and Geochronology), as well those published for the other Cape Verde islands (see below) allow the discussion of some aspects of the archipelago evolution.

The most ancient volcanic rocks outcropping at the Cape Verde Archipelago have been described from the islands of Maio and Santiago, where late Jurassic basalts occur (STILLMAN *et al.*, 1982; GERLACH *et al.*, 1988). However the MORB affinities of these rocks led DE PAEPE *et al.* (1974) to consider them as representing uplifted ocean-floor (see also GERLACH *et al.*, 1988; MILLET *et al.*, 2008).

Considering the age (25.6 Ma) obtained for the Old Eruptive Complex of Sal (TORRES *et al.*, 2002a) and the published geochronological data for the Cape Verde Archipelago (GRIFFITHS *et al.*, 1975; GRUNAU *et al.* 1975;



Fig. 11 – Diagram ¹⁴³Nd/¹⁴⁴Nd vs. ⁸⁷Sr/⁸⁶Sr for Cape Verde archipelago lavas. Mantle component fields and 'Mantle array' from ZINDLER & HART (1986). Square symbols correspond to data in literature (GERLACH *et al.*, 1988; DAVIES *et al.* 1989; JORGENSEN & HOLM, 2002; HOLM *et al.*, 2006; MILLET *et al.*, 2008). Filled squares – Northern Islands (CVN), open squares – Southern Islands (CVS). Black triangles represent new data for Sal island.

MITCHELL *et al.* 1983; CAHEN *et al.*, 1984; TORRES *et al.*, 2002a; PLESNER *et al.*, 2002, MADEIRA *et al.*, 2005; BOSSE *et al.*, 2007; DUPRAT *et al.*, 2007; HOLM *et al.*, 2008; DYHR & HOLM, 2010; FOEKEN *et al.*, 2009; MADEIRA *et al.*, 2010; ANCOCHEA *et al.*, 2010), we conclude that Sal preserves the most ancient emerged rocks of the Cape Verde hotspot. However, considering the evidences for K-Ar age resetting of late Jurassic basalts at 40-50 Ma, which is attributed to hotspot activity (MITCHELL *et al.*, 1983), it is suggested that the Cape Verde plume related magmatism must have started some 25 Ma before the genesis of the first emerged testimonies.

The lack of a well defined island lineament, the proximity to the West African craton and the synchronous tectonic and magmatic evolution of the islands of Maio (Cape Verde) and Fuerteventura (Canary) lead some authors (KING & RITSEMA, 2000; PATRIAT & LABAILS, 2006; KING, 2007) to argue against a mantle plume origin for the Cape Verde archipelago. Indeed, the lack of a well defined rectilinear hotspot track is, at first sight, somewhat surprising. However, it must be remembered that the north-eastward absolute motion of the African plate slowed abruptly 30 Ma ago as a consequence of its collision with Eurasia (SILVER et al., 1998). According to these authors, the African plate drift in the last 10 Ma is lower than 0.8 cm yr⁻¹. If we also take into account that the rotation pole of the African Plate in the last 6 Ma (18.50°N, 17.78°W; POLLITZ, 1991) is relatively close to Cape Verde, we are compelled to accept that a long and well defined hotspot track is not expectable in the Cape Verde region, despite of the long history of volcanic activity.

It should be mentioned that (when considered as a whole) the Cape Verde islands display a very complex age distribution pattern; there is, however, a tendency for east-west age decrease. This is shown by a comparison of the ages of basal complexes of the eastern islands of Sal (25.6 Ma; TORRES *et al.*, 2002a) and Maio (21.1Ma; GRUNAU *et al.*, 1975) with the ages of similar complexes in the western islands of the northern (Santo Antão: 7.6 Ma; PLESNER *et al.*, 2002) and southern islands (Brava: 2.90 Ma; MADEIRA *et al.*, 2010) alignments.

When considered separately, the Southern Cape Verde islands present a clear south-western age decreasing trend, which is primarily suggested by morphological criteria. In fact, Maio Island located on the northeast end of the alignment presents much more eroded landforms than the other Southern Islands (e.g. SERRALHEIRO *et al.*, 1970; MADEIRA *et al.*, 2008). On the other hand the

isotopic ages obtained for Brava (MADEIRA et al., 2010), the fact that the great majority of the exposed volcanics at Fogo are younger than 130 ka (FOEKEN et al., 2009), the occurrence at Fogo of more than 20 eruptions since the discovery and settlement of the island some 550 years ago (TORRES et al., 1997, 1998a), and the occurrence to the southwest of Brava of the Cadamosto seamount, a seismically active 3 km tall volcanic edifice (GREVE-MEYER et al., 2009), suggest that we are faced with a hotspot track resulting from lithosphere migration over a fixed mantle plume, which would now be centred underneath the Fogo/Brava/Cadamosto region (e.g. TORRES et al., 1998b; GREVEMEYER et al., 2009). On the other hand the location of the Northern Islands has been considered as consequence of translithospheric fractures constraining the location of island buildings (TORRES et al., 1998b). Large scale fault-controlled magma transport for more than 3000 km was clearly demonstrated in the Ferrar Large Igneous Province (ELLIOT et al., 1999) conferring feasibility to the above mentioned hypothesis, which may imply "horizontal" magma transport of some hundred km from the upward plume batches of magmas to the local(s) of eruption.

Alternatively, HOLM *et al.* (2008) consider the plume presently centred on a region located Northwest of Sal Island, close to the places where the depth, geoid and heat flow anomalies reach its maxima. Following these authors, channelling by erosion in a swell root by different ponding batches of buoyant plume mantle would explain the spatial and age distribution of the islands. This is clearly an open question.

Notwithstanding that the Cape Verde plume is characterized by a plume flux which is only 9 to 18% of the values reported to Hawaii and Canary islands, respectively (see SCHUBERT *et al.*, 2001; HOLM *et al.*, 2008), it is significant that the Cape Verde Rise constitutes the largest reported oceanic swell (e.g. LODGE AND HELFFRICH, 2006). The above mentioned semi-stationary position of the lithosphere and a long-term magmatic history (during at least 26 Ma) explains how a weak plume produced such a prominent swell (see also MATA *et al.*, 2010).

The data obtained by TORRES *et al.* (2002a) for Sal, allow the conclusion that the outcropping volcanics cover a significant time span (≈ 26 Ma) with two important volcanic hiatus as long as ≈ 3 and ≈ 4 Ma. The volcanism has been inactive since 0.6 Ma ago. The same authors obtained an age 16.3 Ma for an essexitic rock considered to be part of the Old Eruptive Complex. Considering the age determined for the Main Eruptive Formation (15.8 Ma), those plutonic rocks inside the Old Eruptive Complex can be substantially younger than the enclosing rocks, being coeval of volcanics erupted during later stages of the volcanostratigraphic evolution of the island. However, it should be emphasized that an age resetting can not be completely discarded for those essexites.

7.2 Petrogenesis

The estimated olivine saturation temperatures (basalts: 1266 to 1326°C; nephelinites: 1324 to 1363°C; see 5.2) are indicative that Cape Verde primitive magmas were hotter than the expected temperature for the asthenosphere (\approx 1280°C; e.g. MCKENZIE & BICKLE, 1988; WWATSON, 1993). This allows consider that they were originated from a thermally driven mantle plume (see also PUTIRKA, 2008). This percept is endorsed by the estimated temperature of magmatic segregation calculated using the lavas major element composition and the methodology proposed by ALBARÈDE (1992), which are also above 1280°C (see below).

As depicted by the TAS diagram (Fig. 6) Sal lavas are characterized by a limited range of compositions, only basic and ultrabasic rocks being represented. Indeed, intermediate lavas are completely absent, clearly contrasting with that described for other Cape Verde islands where phonolites are volumetrically important (Brava: MOURÃO et al., 2010, MADEIRA et al., 2010; Boavista: DHYR & HOLM, 2010; Santo Antão: PLESNER et al., 2002; HOLM et al., 2006) (see also BEBIANO, 1932). This feature demonstrates that magmatic differentiation processes played a significantly less important role on the chemical variability of the erupted Sal magmas, suggesting that magma chambers were ephemeral and/or volumetrically not significant. This hypothesis is supported by the occurrence of ultramafic mantle xenoliths (see 5.1 Petrography) which indicates that magmas carrying them did not pass through a shallow subcaldera magma reservoir (e.g. CLAGUE 1988). At Sal Island intermediate rocks are only represented as minor volumetric intrusions (Old Eruptive Complex) and by some phonolitic and trachytic dykes cutting the volcanic sequence.

An insight on the genetic processes of the Sal Island magmas is offered by the study of rocks representative of "nearly primary" magmas (Mg# > 58; Ni > 150 ppm). These highly enriched primitive lavas are characterized

by low Yb (1.12 to 2.53 ppm) and Lu (0.14 to 0.30 ppm) contents and by strong fractionation between light and heavy REE [(La/Yb)n = 8.84 to 36.34), characteristics which are indicative of a source characterized by the presence of residual garnet.

The K negative anomalies portrayed by melilitic/nephelinitic rocks have been described in HIMU--type lavas and considered to reflect the K depletion of the subducting oceanic crust occurred during dehydration at subduction zones (LASSITER, 2004). However, for Sal lavas, the increase of Ce/K along with Ce (a proxy of the partial melting degree) indicates a higher compatibility for K than Ce, in opposition to that theoretically expected from melting of sources without K-bearing minerals. This is strongly suggestive that Sal magmas equilibrated with a K-rich residual phase (amphibole and/or phlogopite) capable to preferentially retain K relatively to the adjacent LILE and HFSE (see FRANCIS AND LUDDEN 1995). Considering that the degree of incompatibility of trace elements during partial melting can be inferred from the interception of regression lines in binary diagrams (HANSON, 1989) and that, for Sal $D_K \approx D_{Sr} > D_{Rb} > D_{Ba}$, amphibole is probably the dominant K-bearing residual phase (e.g. $D_{K}^{amph} \approx 1.0$, $D_{Rb}^{amph} \approx 0.3$; $D_{K}^{phlo} \approx 4.0$, $D_{Rb}^{phlo} \approx 6.0$; $D_{Rb}^{amph} \approx 6.0$, $D_{Ba}^{amph} \approx 0.5$; $D_{Ba}^{phlo} \approx 2.1$; $D_{K/Sr}^{mica/liq} \ge 10$ e $D_{Sr/Ba}^{mica/liq} \le 0.1$; see ADAM et al., 1993; LA TOURRETE et al., 1995; FOLEY et al., 1996; Späth et al. 2001).

Accepting that Yb contents are roughly invariant during partial melting of a garnet-bearing mantle source (\approx two times that of chondrites; CLAGUE & FREY, 1982) estimates of melting percentages during Sal magma(s) genesis may be obtained by the inverse approach of MINSTER & ALLÈGRE (1978). This approach substantiates low, but variable, degrees of partial melting, from 1.2 to 4.8 % for melilite/nephelinite magmas to 6.1 to 8.8 % for basaltic magmas (in agreement with experimental data obtained for melting of garnet lherzolite lithologies; e.g. GUDFINNSSON & PRESNALL, 2005).

Considering that garnet is stable in mantle peridotites at depths in excess of 60 km, the thickness of the Cape Verde lithosphere (≈ 83 km; CAZENAVE *et al.*, 1988) and the large abundance of rocks presumably generated by very low degrees of partial melting, we consider that Sal magmas were generated by adiabatic melting of the ascending mantle plume, with the thick lithosphere lid limiting the minimum depth of melting and, consequently, also the amount of magma generated (see WATSON & MCKENZIE, 1991). However, it should be emphasized that amphibole is not stable at temperatures above 1150°C (e.g. WALLACE & GREEN, 1991), i.e., at temperatures prevailing in the asthenosphere or in mantle plumes. Thus, the evidence (presented above) for equilibration of the most undersaturated Sal magmas with amphibole implies the interaction of plume magmas with the lithospheric mantle (see CLASS & GOLDSTEIN, 1997; MATA *et al.*, 1998; SPÄTH *et al.*, 2001). The origin of such hydrated magmas in the lithosphere can be considered as the result of metasomatism, which created low solidus and highly enriched lithospheric domains. In Cape Verde the metasomatic agent has been ascribed either to kimberlitic (Sal) or carbonatitic (Santiago) melts based on the detailed study of mantle xenoliths and of lavas (BONADIMAN *et al.*, 2005; MARTINS *et al.*, 2010).

Estimates of temperature and pressure of magmatic segregation using the methods proposed by ALBARÈDE (1992) and GHIORSO *et al.* (1983) give 1309 to 1422 °C and 1.8 to 2.7 GPa (\approx 54 to 84 km) for basaltic magmas and 1412 to 1578 °C and 3.3 to 4.9 GPa (\approx 102 to 152 km) for foiditic magmas. The obtained sub-lithospheric depths of segregation for the majority of magmas support their origin by adiabatic melting of an ascending plume and suggest that the major element composition of plume magmas was not significantly affected by the interaction with lithosphere.

The significant radiogenic isotope variability of Cape Verde magmas has been explained by variable mixing of HIMU, DMM, EM1 and a lower mantle component. So far, it has been of general consensus that Northern Islands (CVN) magmatic sources reflect the contribution of HIMU, DMM and the lower mantle, the EM1 component being restricted to Southern Islands (CVS) sources (GERLACH *et al.*, 1988; DAVIES *et al.*, 1989; KOKFELT *et al.*, 1998; JØRGENSEN & HOLM, 2002; DOUCELANCE *et al.*, 2003; ESCRIG *et al.*, 2005; HOLM *et al.*, 2006; MILLET *et al.*, 2008; BARKER *et al.* 2009a, b; MARTINS *et al.*, 2010).

Cape Verde EM1-type signatures have been explained by the occurrence in the upper mantle of a megalith(s) of subcontinental lithospheric mantle (SCLM) probably delaminated during the opening of Atlantic ocean. A SCLM origin for the EM1 fingerprints emerged from the geochemical study of lavas (e.g. GERLACH *et al.*, 1988; DOUCELANCE *et al.*, 2003; MILLET *et al.*, 2008; MARTINS *et al.*, 2010) and receive support from a recent geophysical study which imaged a high velocity zone beneath Cape Verde (BEGG *et al.*, 2009). Its P. TORRES; L. C. SILVA; J. MUNHÁ; R. CALDEIRA; J. MATA & C. TASSINARI

presence beneath Sal, one of the Northern Islands, has been inferred from the study of xenoliths. Indeed, BONADIMAN *et al.* (2005) studied spinel-peridotite xenoliths included in Sal lavas and found evidence for the occurrence of kimberlite-like metasomatism, which was interpreted as caused by the melting of a SCLM domain. More recently, COLTORTI *et al.* (2010) presented the results of in situ Os isotope analyses of sulfide grains from the xenoliths. They found Re-depletion model ages mainly ranging from Neoproterozoic to Archean, i.e. compatible with the tectonic history of the western margin of the West African Craton and the corresponding continental margin of Brazil.

Our Sr and Nd isotopic data give some support to these studies suggesting that the presence of SCLM is not restricted to the mantle portions underneath the Southern Islands. Indeed, two of the analyzed Sal samples are characterized by ¹⁴³Nd/¹⁴⁴Nd neatly lower (0.512770 – 0.512771) than those characterizing CVN or the HIMU component, implying the presence of a more enriched (EM1 ?) component in their source (Fig. 11). This percept is also supported by the fact that these two low ¹⁴³Nd/¹⁴⁴Nd samples are also characterized by Ba/Nb > 10.3 and Rb/Nb > 0.72 clearly above the values usually reported for lavas with strong HIMU and/or DMM affinities, but compatible with a contribution of the EM1 component (Weaver, 1991; Willbold & Stracke, 2006; 2010).

Thus, our study demonstrating the presence of EM1-type signatures in lavas from Sal, supports that subcontinental lithospheric megaliths, usually considered to be the responsible by the chemical signatures of Southern Cape Verde islands, also exist on the oceanic mantle beneath, at least, one of the Northern Islands.

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